

Published in final edited form as:

Sci Signal. ; 1(28): re7. doi:10.1126/scisignal.128re7.

Ethanol's Molecular Targets

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Abstract

Ethanol produces a wide variety of behavioral and physiological effects in the body, but exactly how it acts to produce these effects is still poorly understood. Although ethanol was long believed to act nonspecifically through the disordering of lipids in cell membranes, proteins are at the core of most current theories of its mechanisms of action. Although ethanol affects various biochemical processes such as neurotransmitter release, enzyme function, and ion channel kinetics, we are only beginning to understand the specific molecular sites to which ethanol molecules bind to produce these myriad effects. For most effects of ethanol characterized thus far, it is unknown whether the protein whose function is being studied actually binds ethanol, or if alcohol is instead binding to another protein that then indirectly affects the functioning of the protein being studied. In this Review, we describe criteria that should be considered when identifying alcohol binding sites and highlight a number of proteins for which there exists considerable molecular-level evidence for distinct ethanol binding sites.

Introduction

Humans have practiced the art of fermentation for millennia, observing the many actions of ethanol on physiology and behavior in the process. Despite our familiarity with ethanol, we have remarkably little insight into the mechanisms by which it reduces inhibitions and anxiety, nor do we know much about how it produces signs of more severe intoxication. One reason for the difficulty in identifying ethanol receptors is the low binding affinity of ethanol to proteins. This is reflected in the high concentrations required to produce physiological effects. For example, the blood concentration considered legally impairing is 0.08% v/v, which is about 17 mM, and the anesthetic concentration is about 190 mM (1). Ethanol distributes into total body water, so that alcohol concentrations in all organs and cellular compartments are similar to the blood concentration. (One exception is the concentration in the mouth or stomach after ingestion of alcoholic beverages: It is sobering to note that wine, at 12% v/v, has an ethanol concentration of more than 2000 mM!) The high concentrations of ethanol required to produce behavioral effects led to early proposals that ethanol (and other volatile intoxicant anesthetics such as nitrous oxide and diethyl ether) act by perturbing membrane lipids. These early hypotheses have been challenged by evidence supporting proteins as the sites of action (2). This Review focuses on proteins that are believed to have discrete and identified binding sites for alcohols and are affected by ethanol concentrations in the range of 15 to 150 mM. Possible mechanisms of action of alcohols on these proteins are presented.

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Molecular Targets Versus Implicated Molecules

Treatment of cells or animals with ethanol alters the expression and function of many different proteins, and manipulation of many different genes alters alcohol action (3-5). For example, more than 100 different null mutant mice, lacking single specific proteins, are known to exhibit changes in sensitivity to one or more actions of alcohol (6). However, these knockout approaches do not necessarily implicate any of these proteins as direct targets of ethanol action. In addition, gene mapping studies have identified chromosomal regions bearing genes that alter behavioral sensitivity to ethanol (7). Genes identified from these studies may be “involved” in ethanol action, but their roles may be downstream from the initial binding sites for alcohol. In addition, many studies have been published in which alcohol is demonstrated to affect a particular protein or biological process, such as neurotransmitter release or enzyme activity, but the issue of the exact molecular site at which alcohol binds to affect function has not been addressed.

Criteria for an Ethanol Molecular Target

To our knowledge, criteria for defining an ethanol site of action (receptor) have not been articulated, and we hope that our present proposal will generate discussion of this area. There are several criteria of increasing stringency:

1. The function of the protein must be altered by exposure to 15 to 150 mM ethanol in vitro. This alteration should be observed in recombinant and native protein under physiological conditions.
2. Mutation of specific amino acids in the protein should provide evidence of a putative binding site. Evidence for this might include covalent labeling of native or engineered amino acids with alcohol analogs (for instance, labeling cysteine residues with alkane thiols) or mutations designed to increase or decrease the size of the alcohol binding pocket, which could prevent the effects of ethanol or change the size of the alcohol molecule that could bind.
3. Manipulation of this protein in vivo (using knockout or knock-in mice) should produce results consistent with a role in alcohol action.
4. Physical structural studies (x-ray or NMR) should demonstrate the presence of ethanol molecule(s) at the proposed sites within the protein.

We next discuss proteins that fulfill some or all of these criteria.

Strongly Supported Ethanol Targets

Enzymes

Although not normally considered an alcohol “receptor,” alcohol dehydrogenase (ADH) binds alcohol at low millimolar concentrations [the Michaelis constant K_m for mammalian ADH is about 1 mM (8)]. The structure of the alcohol binding site is well known for ADHs from several species (9-11). The size of the alcohol molecule that can bind to ADH is determined by the sizes of the amino acids near the active site, and this can vary among alcohol dehydrogenases from different species (12). For instance, ADH from yeast is only able to metabolize small alcohols. The replacement of a large phenylalanine residue in yeast ADH with the smaller residue (leucine) found at the equivalent position in horse ADH markedly broadens its alcohol specificity (12). A key aspect of alcohol-ADH interaction that has not yet been found in other targets is that the binding of ethanol in ADH requires a zinc atom that forms a bond with the hydroxyl group of the alcohol. It is interesting to speculate that a metal cation may be required for high-affinity (low millimolar) interactions of ethanol with other proteins.

Adenylyl cyclase is an enzyme stimulated by heterotrimeric guanine nucleotide-binding proteins (G proteins) to synthesize the second messenger adenosine 3',5'-monophosphate (cAMP) from adenosine triphosphate (ATP). Different adenylyl cyclase isozymes show different sensitivities to ethanol enhancement of cAMP production: AC3 is completely insensitive; AC2, AC5, and AC6 show moderate sensitivity; and the AC7 isozyme is very sensitive, showing 2 to 3 times as much stimulation of adenylyl cyclase activity by ethanol as the moderately sensitive isoforms (13). Adenylyl cyclase chimeras generated between the AC2 (or AC3) and AC7 isoforms enabled the identification of two "ethanol-responsive domains" that determined the alcohol sensitivity of the chimeric enzyme (14). The first is a 28-amino acid portion of the N-terminal C_{1a} domain, whereas the second falls within the 140-amino acid C-terminal domain. Within this second ethanol-responsive domain is an amino acid sequence quite similar to the alcohol-binding site identified within the *Drosophila* LUSH protein (see below).

Ion channels

Presynaptic cells release neurotransmitters that excite or inhibit postsynaptic cells, depending on which neurotransmitter is released and to what postsynaptic receptors it binds. Ligand-gated ion channels (LGICs) are among the postsynaptic receptors that contain integral ion channels that open in response to neurotransmitter binding. Several LGICs are affected by 30 to 100 mM ethanol and by lower concentrations of longer-chain alcohols (15,16). With respect to behavioral effects, probably the most relevant of the neuronal actions of ethanol on ion channels are its inhibition of *N*-methyl-D-aspartate (NMDA)-type glutamate receptors and its enhancement of γ -aminobutyric acid type A (GABA_A) and glycine receptor function. In addition, *n*-alcohols inhibit the phylogenetically related GABA_C receptors (17), and short-chain alcohols enhance, whereas long-chain alcohols inhibit, neuronal nicotinic receptors (18,19).

Molecular exploration of the alcohol sites in NMDA receptors is in the early stages (20), but there are many structure-function studies of alcohol's actions on GABA_A and glycine receptors. In many of these studies, single or double amino acid mutations were introduced into the target proteins in an attempt to eliminate the actions of alcohols and thereby infer a site of action. Initial studies (21,22) indicated that amino acids in transmembrane (TM) regions 2 and 3 are important for alcohol modulation of glycine receptor function, but residues in TM 1 and 4 may also play a role (23-25). These findings suggested that a cavity exists among the TM regions that may accommodate alcohols. This idea was supported by results of experiments in which changes in the volumes of amino acids lining the putative cavity also changed the size of alcohol that could fit and thus modulate function (22). This site has been estimated to be 250 to 370 Å³ in GABA_A receptors (26). The lack of x-ray crystallographic structural information left open the critical question of whether the implicated amino acids on each of the TM regions were properly oriented (vertically and horizontally in the membrane) to form a cavity. Introduction of cysteine residues at these positions indicated that disulfide bonds could form between double site-directed cysteine mutations introduced in TM 2 and 3, as well as 1 and 3 (23,25), indicating that the cysteine mutants were in close proximity to one another and that those amino acids could participate in forming an alcohol-binding pocket among the TM regions.

The questions of whether this cavity contains water under physiological conditions and can accommodate small alcohols were addressed using alcohol analogs—namely, alkanethiols and alkaneme-thanethiosulfonates—which can covalently label cysteines in the presence of water. Application of compounds such as propyl methanethiosulfonate (PMTS) to GABA_A or glycine receptors mutated to cysteine at critical residues produced irreversible changes in receptor function, indicating that the introduced cysteine was located in a water-filled cavity and that

covalent binding of propanethiol into this cavity irreversibly altered receptor function (24,27, 28). Other studies suggest that the volume of this cavity also plays an important role in receptor activation, providing a mechanism for occupation by small molecules to alter channel function (Fig. 1) (29). Mutations replacing small amino acids (such as serine) within the alcohol binding site with large ones (such as tyrosine or phenylalanine) mimic the effects of modulators such as alcohol and volatile anesthetics by left-shifting neurotransmitter concentration-response curves. Moreover, the putative binding site of alcohol to TM 2 residues is located in an ideal position to affect the transmission of a binding signal from the neurotransmitter binding site to the channel gate on the cytoplasmic side of TM 2.

The energy barrier between the resting and open states of LGICs is low, as demonstrated by spontaneous openings in the absence of agonists, especially in the case of single site-directed mutations (26,30,31). As a result, molecular modeling of the resting-to-open transition requires a pathway in which a substantial structural transition of a large pentameric protein can occur without encountering large energy barriers. The molecular dynamics technique of normal mode analysis has been used to search for low-energy pathways for this transition. One attractive pathway is provided when the helices of a four-helical bundle change their helical crossing angles in a collective fashion. This structural change has been described as a “wringing motion” or a “quaternary twist” (32-35). Figure 1 shows an example of preferential stabilization of one substate in a left-twisted helical bundle. In this illustration, ethanol stabilizes a substate in which the increased helical crossing angle creates a small interhelical cavity.

Some recent studies propose an especially high-affinity ethanol binding site on GABA_A receptors that contain a combination of $\alpha 4$ (or $\alpha 6$), $\beta 3$, and δ subunits, rendering them sensitive to concentrations of ethanol as low as 3 mM (36). In addition, this site is proposed to bind the high-affinity benzodiazepine receptor ligand Ro15-4513 that can compete with ethanol for occupancy of its site (37). However, two studies did not find evidence for competition between ethanol and Ro15-4513 (38,39), and this topic remains controversial. A series of associated reviews devoted to the proposal that specific GABA_A receptors display high sensitivity to ethanol provide data from different laboratories either supporting or failing to support such a site of alcohol action (40).

Three potassium channels are modulated by reasonable (<100 mM) concentrations of ethanol: G protein-activated inwardly rectifying channels (GIRK, Kir3x); large-conductance calcium-activated channels (BK, slo-1, KCNMA1); and Shaw2 voltage-gated channels (Kv3.x, KCNX) (41-44). However, the site of ethanol action has been explored only for the Shaw2 channel, where mutations indicate that the intracellular S4-S5 linker is critical for inhibition by alcohols (45). Furthermore, kinetic and structural data suggest that an amphipathic α -helical region of the linker near S5 is the site of alcohol action (45) and that the S6-b segment may contribute another α helix that is important for alcohol binding (46). Thr¹⁰⁷ in a cytosolic loop of the bovine slo-1 channel influences BK channel responses to ethanol, but this residue appears to affect channel function through phosphorylation rather than constituting a binding site for alcohol (47).

Other proteins

Drosophila uses odorant binding proteins to detect ethanol, and one of these proteins, LUSH, is required for avoidance of high, potentially toxic, concentrations of ethanol; LUSH can also bind propanol and butanol, but not longer-chain alcohols. LUSH provides the best-characterized nonenzymatic site of alcohol action. The crystal structure shows a single alcohol binding site in a water-filled cavity that accommodates a single alcohol molecule (Fig. 2) (48). Replacement of multiple water molecules by the alcohol appears to provide favorable entropy for alcohol binding. The cavity lies between α helices, with Thr⁵⁷ and Ser⁵² providing hydrogen bonding with the alcohol; hydrophobic residues likely interact with the methylenes

(49). Occupancy of the cavity is seen at aqueous ethanol concentrations of 30 to 50 mM. In the absence of alcohol, LUSH oscillates among multiple conformational states, and replacement of water by the single alcohol molecule provides conformational stability (50). The key findings from the LUSH structure (a single water-filled cavity between α helices using serine and hydrophobic amino acids to bind alcohol and stabilize a protein conformation) was proposed earlier (23,25) on the basis of extensive mutagenesis of the glycine receptor, as described above.

Apart from producing sedation and motor incoordination, alcohol consumption by pregnant women can also lead to the development of fetal alcohol syndrome (FAS); at least some of the features of FAS are believed to be due to alcohol-mediated inhibition of neuronal adhesion (51), specifically through the L1 cell adhesion molecule (52). A recent report identified specific amino acids forming an alcohol binding pocket between the Ig1 and Ig4 domains of L1 to which azibutanol and azioctanol, photoactivatable analogs of butanol and octanol, could bind (53). Thus far, no antagonist has been identified that specifically blocks the effects of ethanol on any ion channel. However, octanol is an antagonist of ethanol inhibition of L1-mediated cell adhesion (54). In addition, small peptides also antagonize alcohol effects on L1 (55,56). Two peptide fragments, each only 8 or 9 amino acids in length, are neuroprotective at femtomolar concentrations in preventing ethanol-induced teratogenesis and cell death in whole mouse embryo cultures. These peptides also antagonize ethanol inhibition of L1 adhesion (55). Moreover, the dietary administration of one of these peptide fragments reduces the incidence of alcohol-induced ocular teratogenicity (57). These recent findings not only demonstrate the feasibility of searching for peptide antagonists of alcohol actions on proteins, but also show that these peptides can be expected to act with high affinities and might even be able to be administered as therapeutics.

Common Themes Among the Targets

It is possible that the low-affinity binding of small alcohols could be accommodated by many types of protein structures, and our limited knowledge of molecular sites to which alcohols bind will not provide any common theme. However, there is a remarkable convergence in the alcohol-binding sites of the known targets. Many of the putative binding pockets have two or more α helices with amphipathic surfaces, often containing serine or threonine residues. At least some of these cavities normally contain water, and the displacement of water by ethanol is likely critical for the functional changes. The effects of 1-alkanols on these proteins are also limited by chain length, although this limitation varies markedly from LUSH (where only ethanol, propanol, and butanol are effective) to glycine and GABA_A receptors, where alcohols ranging in size from methanol to decanol enhance function.

Relevance of Defining Sites of Ethanol Action

There are two potential applications of a better understanding of molecular sites of alcohol action. First is the possibility of designing alcohol antagonists or partial agonists. Partial agonists for nicotinic and opioid receptors are useful for treating nicotine and opioid dependence (58,59), and it is possible that drugs acting at alcohol binding sites might reduce craving or relapse in alcoholics; the feasibility of therapeutic alcohol antagonists is supported by studies of the L1 adhesion molecule. Second, there is evidence that individuals who have low sensitivity to alcohol intoxication are prone to develop alcoholism (60). It is possible that genetic variation in genes coding for alcohol binding sites may be important in alcoholism. The results of the recent Human Genome Project could soon allow prediction of an individual's propensity for alcoholism, resulting in proactive programs of education and intervention.

Future Directions

Which proteins, beyond those discussed above, have cavities where alcohol can bind and consequently alter protein function? In particular, where and how does ethanol act to produce the threshold intoxication seen at low millimolar (e.g., 10 mM) concentrations? With the exception of ADH, L1, and possibly a couple of specific GABA_A receptor subtypes, the targets discussed above require 50 to 150 mM ethanol for substantial changes in protein function, at least when these targets are studied *in vitro*. Perhaps intracellular modulators affect alcohol sensitivity and this sensitivity is decreased in experiments conducted *in vitro* in the absence of these modulators. It is also noteworthy that ADH achieves the high-affinity binding of alcohol by using zinc to interact with the hydroxyl group of ethanol. Are there non-enzymatic sites in brain proteins that use a similar strategy?

Acknowledgements

We thank C. Borghese for helpful comments, advice, and insight. Supported by NIH grants R01AA006399 and P01GM047818 (R.A.H.), R01AA013378 (J.R.T.), and R01AA11525 (S.J.M.).

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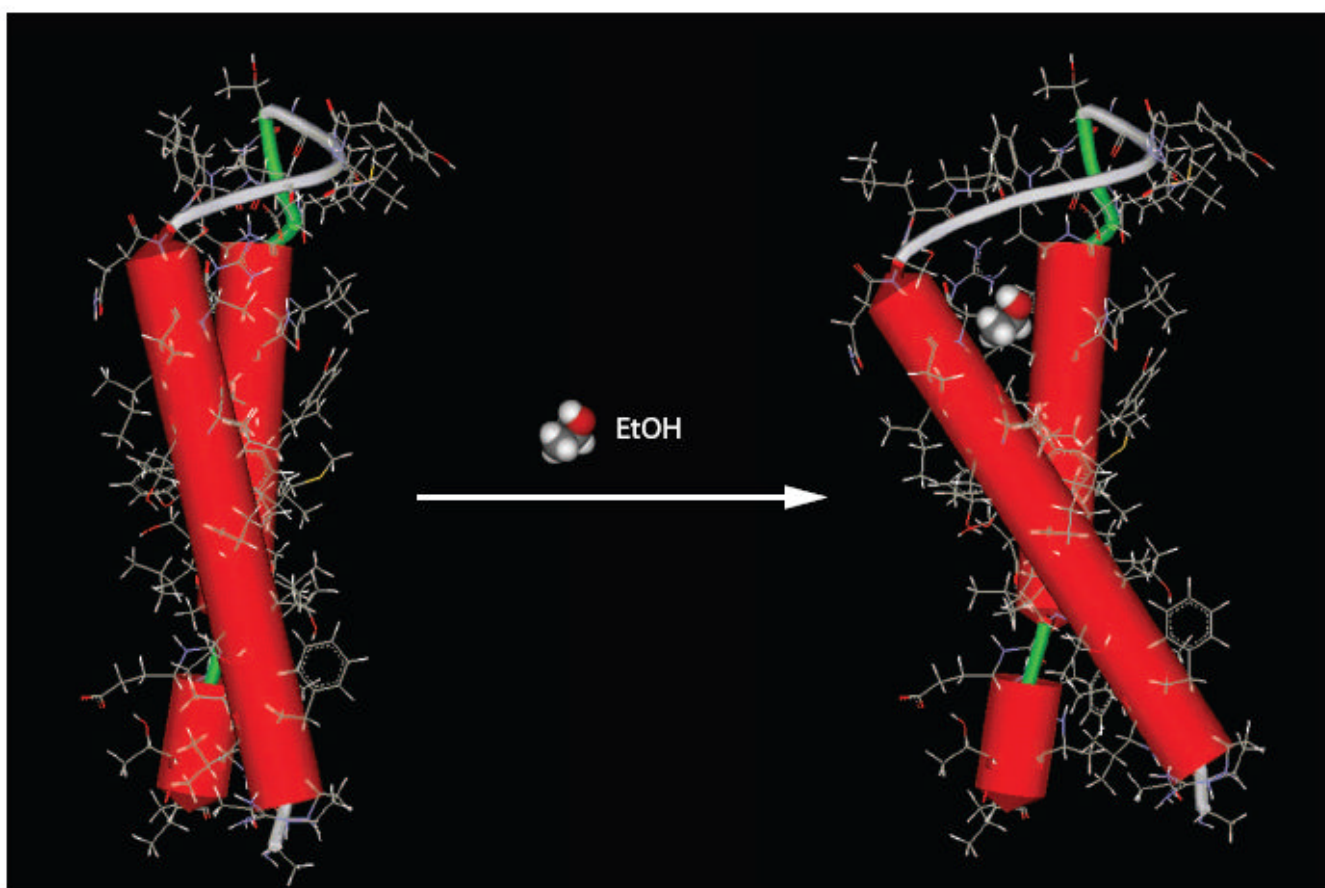


Fig. 1. Preferential stabilization of a substate by ethanol binding. In this illustration, two α helices connected by a short linking loop represent the TM2-TM3 helices found in one subunit of a GABA or Gly receptor. **(Left)** One substate with a narrow left crossing angle between the helices. **(Right)** A second vibrational substate with a wider left crossing angle is stabilized by binding a molecule of ethanol in a small cavity created by the intersection of the two helices. If the model in the right panel represents the tertiary structure of the open state of the ion channel in a LGIC, binding of ethanol would favor the transition to the open state.

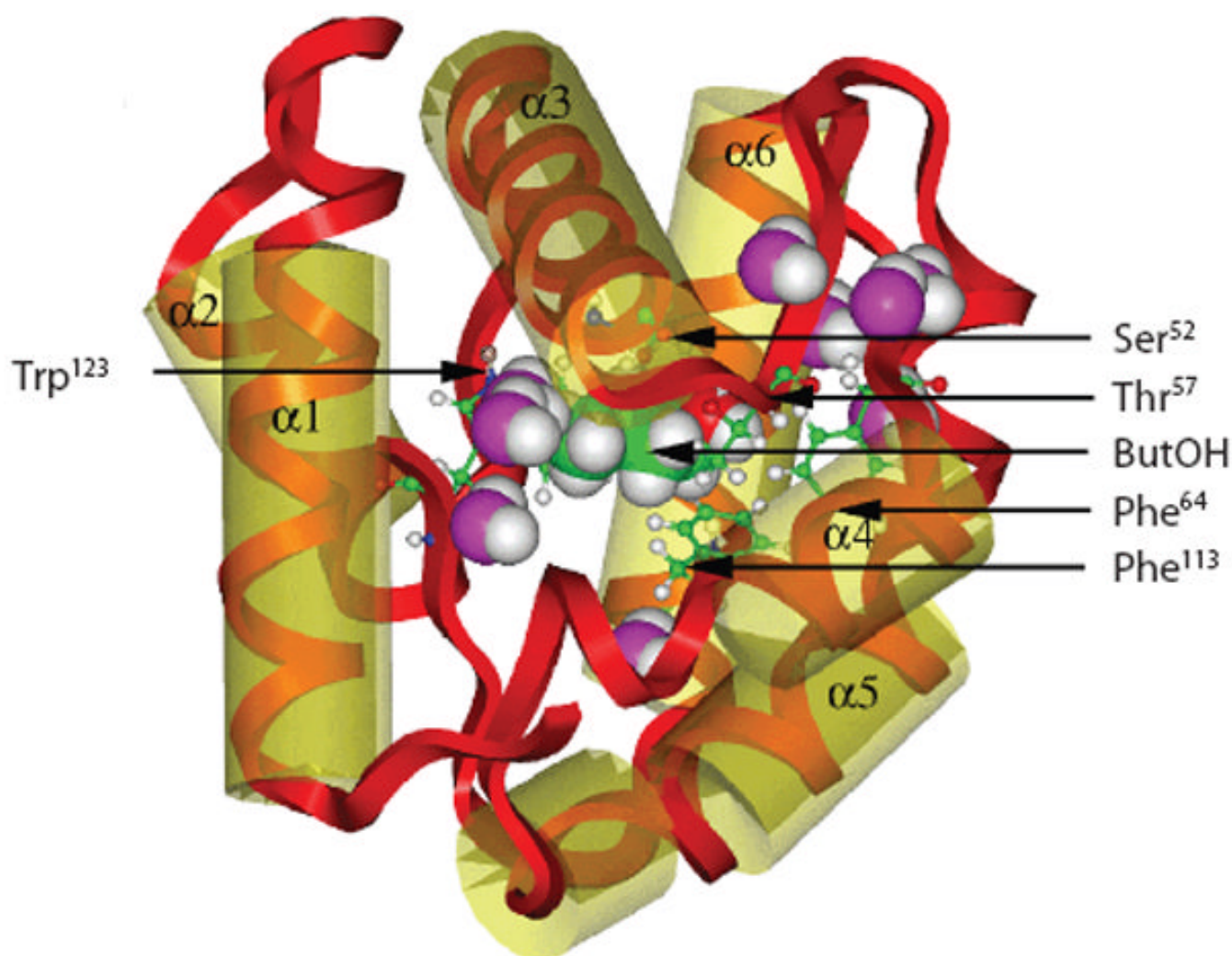


Fig. 2.

Butanol at the interhelical binding site in LUSH. [The PDB file of butanol in the binding site of LUSH (1OOH), complete with 329 water molecules, is from Kruse *et al.* (48).] A subset of nine water molecules within 10 Å of the oxygen atom of butanol (ButOH) was identified with Insight II (Accelrys, San Diego, California); the subset is highlighted using pink van der Waals surfaces. The five amino acid residues identified in the binding pocket of LUSH by Kruse *et al.* (48) are rendered with stick surfaces showing carbon, green; oxygen, red; nitrogen, blue; and hydrogen, white.